



## DNA-Templated Silver Nanoclusters as a Label-Free Optical Sensor for Cisplatin Detection

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### Abstract

Cisplatin or *cis*-diamminedichloroplatinum (II), is a potent anti-cancer drug widely applied to treat various types of cancer. Unfortunately, it also has various negative effects, such as neuro and renal toxicity. Therefore, it is of critical importance to avoid cisplatin overdosage in patients and to avoid cisplatin contamination in chemotherapy facilities. Conventionally, the detection of cisplatin in relevant samples of interest can be accomplished using various techniques, such as high-performance liquid chromatography (HPLC); however, these methods are often expensive and require skilled operators. To address the issue, this study proposes a simple fluorescent biosensor based on DNA-templated silver nanocluster (AgNC), with the specific binding between DNA and silver ions and acting as a reducing agent to convert silver ions into silver nanoclusters. Under the 570 nm excitation light, the nanocluster emits bright fluorescence (peaked at 620 nm) partially due to the enhancement from nearby guanine bases on the DNA-template. However, the fluorescence intensity is quenched by cisplatin likely because of cisplatin – guanine bonding. Therefore, cisplatin can be easily quantified by measuring the decrease in AgNC fluorescence emission. Under optimal conditions, the biosensor demonstrates a good linear range of detection from 5.0 to 40.0  $\mu$ M, a low limit of detection (LOD) of 2.7  $\mu$ M and a limit of quantitation (LOQ) of 9.0  $\mu$ M. With further development, the developed biosensor should be able to measure cisplatin in samples of interest, such as biological fluids from patients undergoing chemotherapy or water samples from wastewater treatment plants in hospitals.

**Keywords:** *cisplatin, biosensor, DNA-templated AgNC, silver nanocluster*

### 1. Introduction

*Cis*-diamminedichloroplatinum (II), or cisplatin is a potent anti-cancer drug that is widely used to treat various types of cancer, including lymphomas, sarcomas, skin, ovarian, testicular, bladder and lung cancer (Christie & Tansey, 2007; Jung & Lippard, 2007; Dasari & Bernard Tchounwou, 2014). Once entering cancer cells, cisplatin molecules turn into a more reactive diaqua form that bonds to the N7 position of one or two purine bases, with high preference for consecutive guanines (Wang & Lippard, 2005; Wang et al., 2013). The resulting DNA-cisplatin adducts interfere with DNA replication and transcription, ultimately leading to cell death (Wang & Lippard, 2005). However, cisplatin does not distinguish between cancerous and normal cells. Therefore, it can cause severe negative side effects, such as neuro and renal toxicity (Cornelison & Reed, 1993; Smoorenburg et al., 1999; Dasari & Bernard Tchounwou, 2014). Ironically, cisplatin is classified as a type II carcinogenic substance (National Toxicology Program, 2021).

Unintentional cisplatin exposure may be prevented by closely monitoring cisplatin contamination in chemotherapy workplaces, in patient's body fluid, especially urine samples or in wastewater treatment facilities within hospitals. Conventional techniques for cisplatin detection include liquid chromatography mass spectrometry (LC-MS) (Shaik et al., 2017), inductively coupled plasma mass spectrometry (ICP-MS) (Brouwers



et al., 2008), and high-performance liquid chromatography (HPLC) (Toro-Córdova et al., 2016). Despite being highly specific and sensitive, these techniques are expensive and requires well-trained personnels to operate. To address the problems, several research groups developed cisplatin fluorescent biosensors as simpler and cheaper alternatives (Yang et al., 2014; Jantararat et al., 2021). However, with a few exceptions, they are still based on organic fluorescent dyes that are sensitive to ambient light and often require expensive labeling.

Fluorescent nanostructures, such as quantum dots, QDs offer intriguing alternatives due to their high robustness, high fluorescence quantum yields, and extended shelf life (Farzin & Abdoos, 2021). Despite tremendous efforts to apply greener approaches in recent years, many QDs syntheses still involve toxic reagents and tedious processes (Liu et al., 2009, Dharmo et al., 2022). On the other hand, metallic nanoclusters (NCs), such as silver NCs, are gaining widespread support due to their ease of synthesis and requires lesser or non-toxic substances required during synthesis. Even though AgNCs can be synthesized as free clusters under suitable reducing conditions, DNA containing certain sequence rich in cytosine offers a stabilizing template for AgNC formation (Petty et al., 2004, Song et al., 2019). The resulting DNA-templated AgNCs shows tunable fluorescence excitation and emission peaks and high photo-stability (Xu et al., 2020). In addition, they are sensitive to microenvironments, such as ions (Lee et al., 2014). Accordingly, they are utilized as biosensors for detecting small molecules, DNA, or proteins (Sharma et al., 2010, Sharma et al., 2011, Liu et al., 2013; Zheng et al., 2015).

What interests us is that the fluorescence emission of DNA-templated AgNCs can be enhanced by nearby guanine bases (Li et al., 2020; He et al., 2021; Fredrick et al., 2023). They proposed that this enhancement is due to electron transfer between the nearby guanine bases and AgNCs. As a result, we hypothesize that the AgNC emission enhancement will be suppressed in the presence of cisplatin-DNA adduct formation after cisplatin-guanine's N7 binding allowing the DNA-templated AgNCs to be used as a simple cisplatin biosensor.

## 2. Objectives

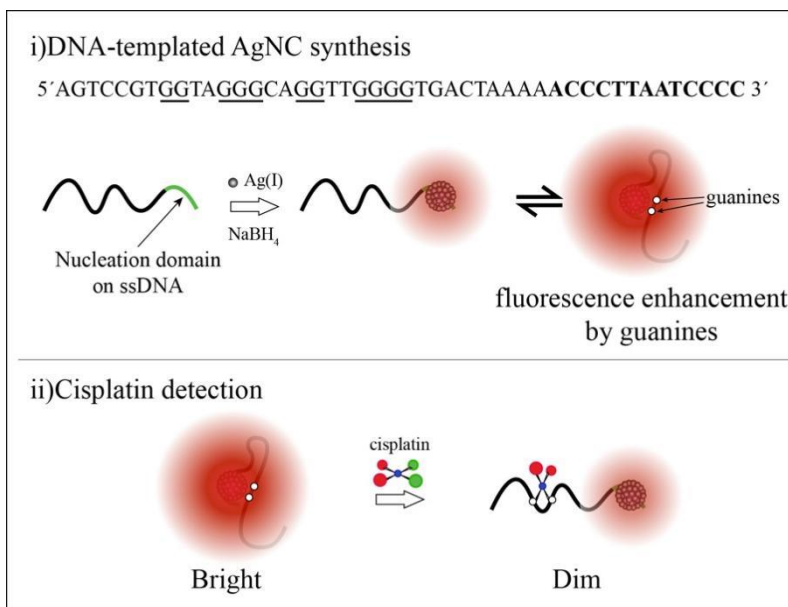
To develop a simple and economical optical biosensor for cisplatin detection.

## 3. Materials and Methods

### 3.1 DNA-templated AgNCs synthesis and the proposed sensing mechanism

The DNA sequence named Ag4 (shown in Step 1 of Figure 1) is derived directly from a prior work (Sharma et al., 2010). It contains a nucleation domain (bolded letters) for  $\text{Ag}^+$  ions binding and AgNC formation and multiple guanines for AgNC fluorescence enhancement and cisplatin binding (underlined letters). The nanoclusters synthesis is done following a protocol modified from a prior work by Sharma and co-workers (Sharma et al., 2012). Briefly,  $3.0 \mu\text{M}$  of the DNA template was mixed with  $18.0 \mu\text{M}$   $\text{AgNO}_3$  in  $10.0 \text{ mM}$  HEPES buffer pH 8.2 (approximately the highest for HEPES) to stabilize  $\text{NaBH}_4$ , and kept in darkness for 10 minutes to allow  $\text{Ag}^+$  and DNA binding. Then  $18.0 \mu\text{M}$   $\text{NaBH}_4$  was added to reduce  $\text{Ag}^+$  to  $\text{Ag}^0$  and facilitate the AgNC formation on the DNA template. Next, the mixture was vigorously mixed by a vortex mixture for 1 minute and incubated in darkness for 4 hours at  $4^\circ\text{C}$ . The DNA-template AgNC was then kept at  $4^\circ\text{C}$  until use.

For the proposed cisplatin sensing as shown in Step 2 of Figure 1, in the absence of cisplatin, the DNA-templated AgNCs emits an intense fluorescence possibly due to electron transfer from nearby guanine bases, as proposed in a prior work (Zoughi et al., 2024). In the presence of cisplatin, however; the cisplatin-guanine complex formation greatly alters the efficiency of electron transfer causing the reduction of AgNC fluorescence intensity. Consequently, one may quantify cisplatin concentration simply by measuring the reduction in AgNC fluorescence.



**Figure 1** DNA-templated AgNC synthesis and the proposed sensing mechanism

### 3.2 Chemicals and DNA

HEPES (4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid), AgNO<sub>3</sub>, and NaBH<sub>4</sub> were purchased from Sigma-Aldrich (Darmstadt, Germany), while NaOH and NaCl were sourced from Merck (either in Darmstadt, Germany or Copenhagen, Denmark). The DI water having a resistivity of 18.2 MΩ.cm was from a Milli-Q system (Merck, Germany). The Ag<sub>4</sub> DNA sequence (5' AGTCCGTGGTAGGGCAGGTTGGGGTGACTAAAAACCCTTAATCCCC 3') was purchased from Macrogen (Korea). After resuspending in DI water, the DNA sample was divided into aliquots and stored at -20 °C until use.

### 3.3 Standard cisplatin activation and preparation

The preparation of cisplatin in its diaqua form ([Pt(NH<sub>3</sub>)<sub>2</sub>((H<sub>2</sub>O)<sub>2</sub>)<sup>2+</sup>) with the help of Ag<sup>+</sup> to remove extra chloride ions was done following the protocol reported earlier (Jantararat et al., 2021). Briefly, a 5.0 mM cisplatin was dissolved in deionized water and incubated in the dark at 25 °C overnight. Subsequently, 5.0 mM of AgNO<sub>3</sub> was added to provide Ag<sup>+</sup> that prevented the re-association of chloride ions with the diaqua cisplatin. The solution was then incubated in the dark at 25 °C for another 15 hours before centrifugation at 12,000 rpm and 4 °C to remove AgCl. The standard diaqua cisplatin was kept at 4 °C in the dark before use. For the preparation of diaqua cisplatin without Ag<sup>+</sup>, the process was repeated without AgNO<sub>3</sub> added.

### 3.4 Fluorescence Measurements

For each fluorescence emission spectrum measurement, 3.0 μM of freshly prepared DNA-templated AgNC was first mixed with a cisplatin standard solution at the desired concentration in 10.0 mM HEPES buffer at pH 6.8 (approximately the minimum for HEPES) to maintain cisplatin in its diaqua form (Andersson et al., 1994). Subsequently, 120 μL of the mixture was transferred to a micro-cuvette (16.100-Q-10/Z8.5, Starna Cells Inc., USA) for measurement in a benchtop fluorometer (F-2700, Hitachi, Japan) having the excitation wavelength

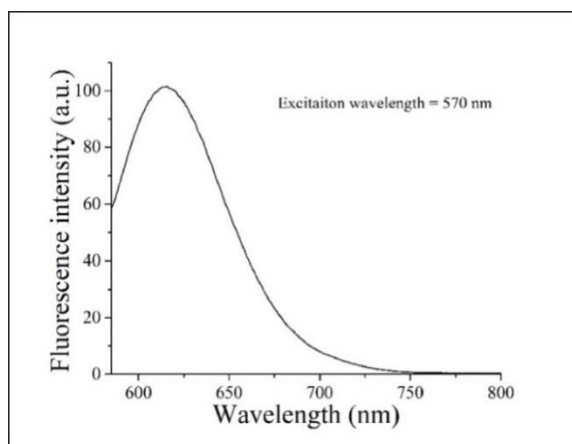


of 570 nm, and emission wavelengths ranging from 585 nm to 800 nm. The excitation and emission slits were set at 5 nm and 10 nm, respectively.

#### 4. Results and Discussion

##### 4.1 Fluorescence Emission from DNA-Templated Silver Nanocluster

Under 570 nm excitation, the solution containing freshly prepared DNA-templated AgNC emits red fluorescence with a peak at 620 nm, as shown in Figure 1. This peak position of the fluorescence emission spectrum is somewhat different from 700 nm reported earlier for the Ag4 DNA template (Sharma et al., 2010), possibly due to the difference in the buffer composition and pH being used. Nevertheless, the fluorescence intensity of the DNA-templated AgNC in our work was still high hinting that it was successfully synthesized. It is worthwhile mentioning that only vigorous shaking over a brief period, as described in section 3.1 is critical because the solutions prepared differently did not fluoresce (data not shown). To eliminate any instrumental effects, the area under the fluorescence emission spectrum in the presence of cisplatin at a given concentration (F) was divided by that of the solution without cisplatin ( $F_0$ ). This  $F/F_0$  ratio was then used as the biosensor's signal in the key parameters optimization and evaluating the linear range and limit of detection.



**Figure 2** Fluorescence emission spectrum of silver nanoclusters formed on Ag4 ssDNA taken from prior work (Sharma et al., 2010)

##### 4.2 Key Parameters Optimization

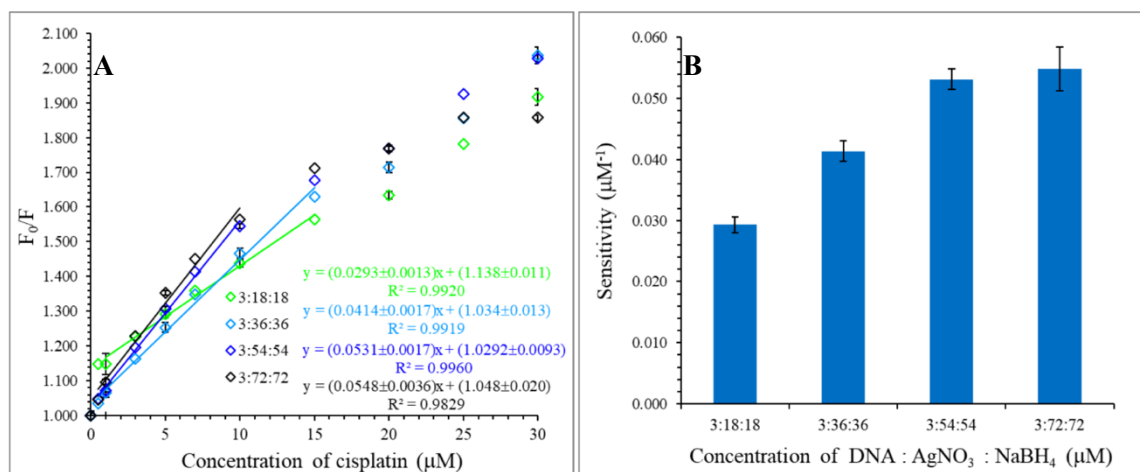
Next, key parameters affecting the performance of the developed DNA-templated AgNC were optimized. During the optimization, one parameter was varied while the rest was fixed, with the experiments were conducted 3 times for each parameter. The criteria for selecting the optimal values were the highest sensitivity, defined as the slope of the calibration graphs, and the shortest analysis time.

###### *The concentrations of template DNA, AgNO<sub>3</sub> and NaBH<sub>4</sub> during AgNC synthesis*

A preliminary test (data not shown) suggested that the AgNO<sub>3</sub>:NaBH<sub>4</sub> ratio should be kept at 1:1 to avoid the problem of excess Ag<sup>+</sup> interfering with sensor performance. Therefore, during the step of optimizing DNA, AgNO<sub>3</sub>, NaBH<sub>4</sub> concentrations, the concentrations of AgNO<sub>3</sub> and NaBH<sub>4</sub> were varied together while the concentration of DNA was kept at 3.0 μM. The results shown in Figure 3 suggested that biosensor sensitivity increases with rising AgNO<sub>3</sub> and NaBH<sub>4</sub> concentrations from 18.0 to 54.0 μM before reaching the plateau. The early increase is likely due to the fact that increasing AgNO<sub>3</sub> and NaBH<sub>4</sub> leads to more formation of AgNC-DNA



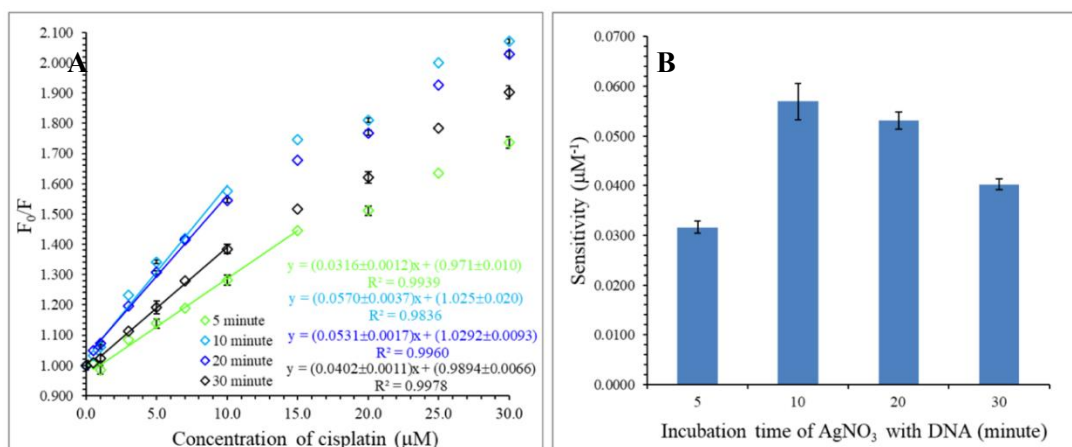
and a better response to cisplatin. However, at high concentration, such as 72.0  $\mu\text{M}$ , the DNA template could be exhausted and no further increase in the sensitivity was observed. To ensure maximum sensitivity and stability, the 3.0:72.0:72.0  $\mu\text{M}$  were chosen as the optimal concentrations.



**Figure 3** Calibration graphs (A) and bar graphs (B) comparing the sensitivities of the sensor at various concentrations of DNA: AgNO<sub>3</sub>: NaBH<sub>4</sub>

#### The incubation time for Ag<sup>+</sup> - DNA template binding

Next, the incubation time of DNA and AgNO<sub>3</sub> prior to mixing with NaBH<sub>4</sub> was optimized. The results presented in Figure 4 indicate that 10 minutes gave the highest sensitivity. We hypothesize that at 5 minutes, the incubation time is too short for Ag<sup>+</sup> to stably bind to the DNA template, while at 20 and 30 minutes (or longer), Ag<sup>+</sup> might start to crosslink two cytosines resulting in unwanted DNA structures. Both of these negative effects could reduce the sensitivity of the sensor. Therefore 10 minutes was chosen as the optimal duration for DNA-templated AgNO<sub>3</sub> incubation time.

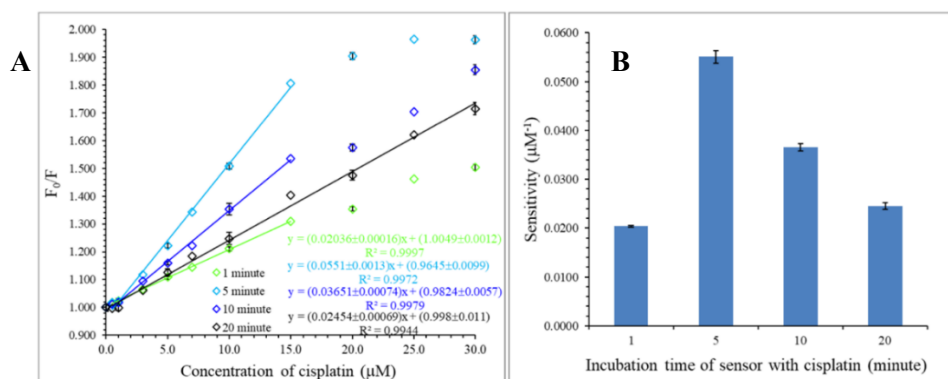


**Figure 4** Calibration graphs (A) and bar graphs (B) comparing the sensitivities of the sensor at various incubation times for the DNA and AgNO<sub>3</sub> mixture



#### The incubation time for cisplatin – DNA-templated AgNC binding

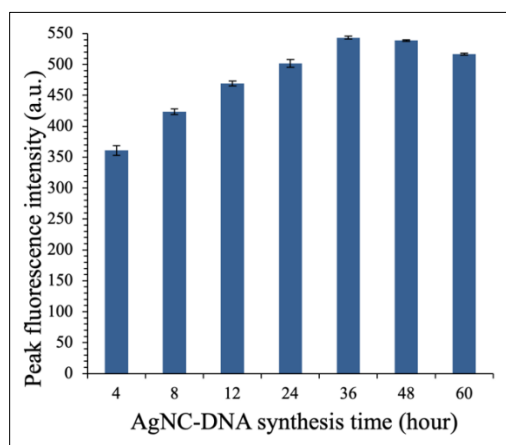
The incubation time between cisplatin and AgNC-DNA was optimized. The results presented in Figure 5 shows that 5 minutes is the optimal value. We hypothesize that at 1 minute incubation time, cisplatin still does not stably bind to the AgNC-DNA probe, while at 10 and 20 minutes (or longer), cisplatin may bind to more than one guanine resulting in a decrease in the inhibition of electron transfer from guanine to AgNC on the DNA. While this hypothesis awaits testing, 5 minutes has been selected as the optimal incubation time.



**Figure 5** Calibration graphs (A) and bar graphs (B) comparing the sensitivities of the sensor at various incubation times for the DNA-templated silver nanocluster and  $\text{AgNO}_3$

#### The DNA-templated AgNC synthesis time

It can be observed from Figure 6 that increasing the incubation time from 4 hours to 36 hours improved the fluorescence intensity emitted by the AgNC-DNA. This improvement is likely because the longer incubation time allows extra AgNCs to form on the DNA template. After 36 hours; however, the peak fluorescence intensity decreases, possibly due to the formation of larger silver nanoparticle since prolonged incubation may allow more silver ions to deposit on the silver nanocluster turning it into a bigger and non-fluorescent silver nanocluster. Therefore, 36 hours was chosen as the optimal incubation time for AgNC-DNA synthesis



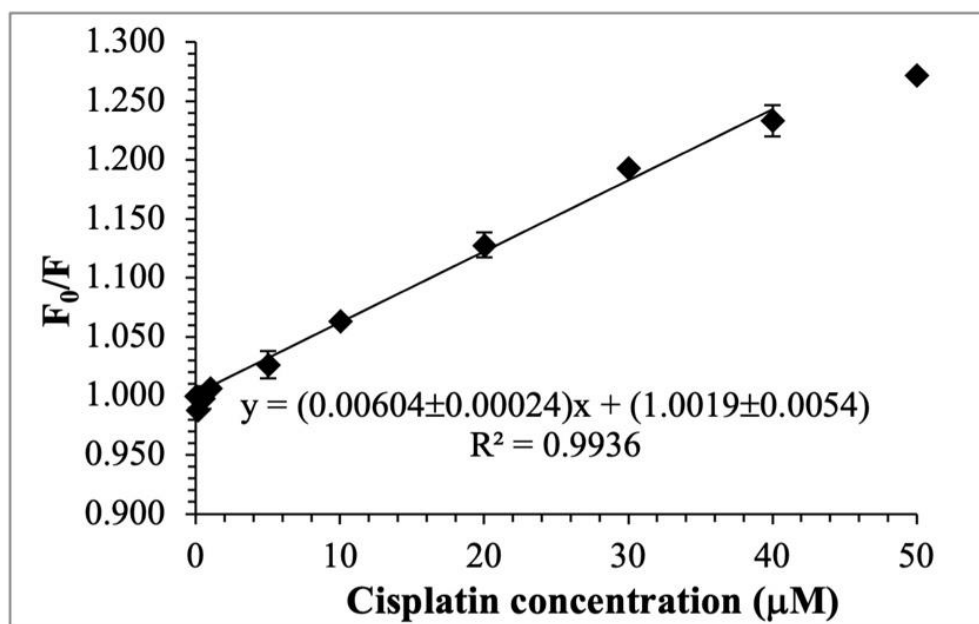
**Figure 6** The effect of AgNC-DNA synthesis time on the peak fluorescence intensity





#### 4.3 The Linear Range and Limits of Detection

After optimization, the linear range and limits of detection were obtained from a calibration graph plotted between  $F/F_0$ , measured under optimal conditions, and the concentration of standard cisplatin solutions. The results presented in Figure 7 indicate a linear range of detection from 5.0 to 40.0  $\mu\text{M}$  from which the coefficient of determination ( $R^2$ ) very close to 1.0000. The limit of detection ( $\text{LOD} = 3S_a/b$ , where  $S_a$  is the standard deviation of the blank response and  $b$  is the slope of the calibration graph (Currie, 1999)) equals 2.7  $\mu\text{M}$  and the limit of quantitation ( $\text{LOQ} = 10S_a/b$ ) equals 9.0  $\mu\text{M}$ . Even though the LOD and LOQ are not lower than some of the previous works (Yang et al., 2014, Jantararat et al., 2021), the merits of the developed DNA-templated AgNC are still high because of its simplicity, low cost, and long excitation and emission wavelengths where interference from matrices tend to be minimal.



**Figure 7** The calibration graph obtained under optimal conditions revealing the linear range of detection from 5.0 to 40.0  $\mu\text{M}$ , with a limit of detection (LOD) of 2.7  $\mu\text{M}$  and with limit of quantitation (LOQ) of 9.0  $\mu\text{M}$

#### 5. Conclusion

In this study, a simple biosensor based on a DNA-templated silver nanocluster (AgNC) was successfully synthesized having a peak excitation wavelength of 570 nm and a peak fluorescence emission wavelength of 620 nm. The biosensor was sensitive to the presence of cisplatin possibly because cisplatin bonding to the DNA template alters the efficiency of electron transfer from guanine bases to the AgNC. Under the optimal sensing conditions, the developed biosensor was able to detect cisplatin in standard solutions with a linear range of detection from 5.0 to 40.0  $\mu\text{M}$ , with a limit of detection (LOD) of 2.7  $\mu\text{M}$  and a limit of quantitation of 9.0  $\mu\text{M}$ . With further development, it is expected that the biosensor will find applications in the detection of cisplatin contamination in various samples, such as urine collected from patients receiving chemotherapy or wastewater treatment plant in hospitals.



## 6. Acknowledgements

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